

# Profiling of Polar Organic Acids in Mouse Muscle Using Ion Chromatography/Mass Spectrometry

Christopher Petucci<sup>1,2</sup>, Andrew V. Zelenin<sup>1,2</sup>, Jeffrey A. Culver<sup>1,2</sup>, Meghan Gabriel<sup>1</sup>, Ken Kirkbride<sup>3</sup>, Terri T. Christison<sup>3</sup>, and Stephen J. Gardell<sup>1,2</sup>

<sup>1</sup>Metabolomics Core Facility, Sanford Burnham Prebys Medical Discovery Institute, 6400 Sanger Road, Orlando, FL 32827

<sup>2</sup>Southeast Center for Integrated Metabolomics (SECIM), Clinical and Translational Science Institute, 2004 Mowry Road, Gainesville, FL 32610

<sup>3</sup>Thermo Fisher Scientific, 1214 Oakmead Parkway, Sunnyvale, CA 94088



## Overview

- Ion chromatography (IC) was evaluated to demonstrate its ability to highly retain and separate polar metabolites that are poorly retained and resolved by reversed-phase chromatography.
- Ion chromatography was coupled with mass spectrometry (IC/MS) to quantitate a variety of polar organic acids in mouse muscle.
- The organic acids examined cover several metabolic pathways including glycolysis, the pentose-phosphate pathway, the TCA cycle, nucleotides, and amino acids.

## Introduction

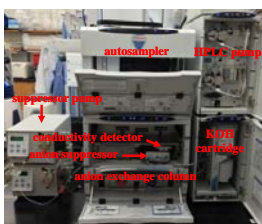
- Organic acids (OAs) are crucial metabolites that play a pivotal role in a host of different metabolic and regulatory pathways.
- We have developed a novel analytical method to quantitate 27 different polar OAs using ion chromatography/triple quadrupole mass spectrometry (IC/MS).
- The method was used to quantitate OA differences in quadriceps muscles from sedentary mice compared to mice that underwent a low intensity, long duration (LILD) or high intensity short duration (HISD) forced treadmill exercise regimen.
- This pilot study has demonstrated that IC/MS is a powerful new tool to separate and quantitate low molecular weight, polar metabolites that are difficult to analyze by other techniques.

## Methods

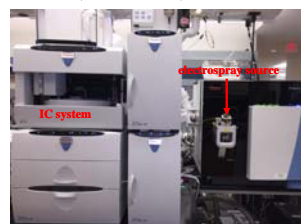
- Frozen quadriceps from mice were lyophilized with subsequent powdering and homogenization of 5 mg of powder in 500  $\mu$ L of 50:50 acetonitrile/0.3% formic acid using a Precellys Evolution homogenizer.
- A 50  $\mu$ L aliquot of homogenate, spiked with <sup>13</sup>C, <sup>2</sup>H, or <sup>15</sup>N internal standards, was derivatized with 50  $\mu$ L of 0.2 M benzylhydroxylamine to stabilize keto-acids.
- The sample was extracted with 1 mL of ethyl acetate followed by drying down 900  $\mu$ L of the organic layer with reconstitution in 100  $\mu$ L of deionized water for IC/MS analysis.
- Calibration curves, spiked with <sup>13</sup>C, <sup>2</sup>H, or <sup>15</sup>N internal standards, (Cambridge Isotope Laboratories) were prepared the same as tissue samples from 0.1-250  $\mu$ M or up to 5000  $\mu$ M for lactic acid.
- A Thermo Scientific ICS-5000 ion chromatography system using a Dionex AS11-HC, 4  $\mu$ m, 2 x 250 mm anion exchange column was used to separate the OAs using a KOH gradient from 5 mM KOH to 100 mM KOH over 11 min.
- The IC was coupled to a Thermo Scientific Quantiva triple quadrupole mass spectrometer to quantitate the OAs by single reaction monitoring via electrospray ionization in the negative ion mode.

## IC/MS Instrumentation

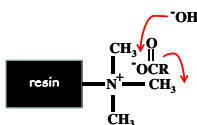
Thermo Scientific ICS-5000+ Reagent Free HPIC System



Thermo ICS-5000+ HPIC/Quantiva Triple Quadrupole MS

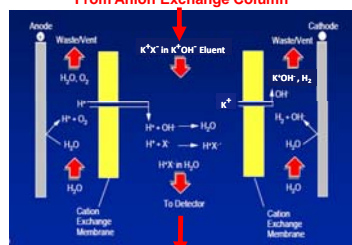


Anion Exchange Stationary Phase and Mechanism



## Anion Suppressor

From Anion Exchange Column



To Electrospray Mass Spectrometer

- The anion suppressor neutralizes the KOH eluent, protonates carboxylic acids (HX), and removes potassium cations.
- Hydroxide and potassium ions can contaminate the conductivity detector. Also, potassium salts are not volatile by electrospray ionization.

## Results

### Precision and Accuracy of Representative Calibrators Over 3 Days

Organic Acid	Nominal Concentration ( $\mu$ M)	Mean Measured Conc. ( $\mu$ M), n = 6	S.D. (n = 6)	% CV	% Accuracy
lactic acid	50	51	2	4	101
	200	201	17	9	100
	500	502	28	6	100
	1000	951	31	3	95
	2000	2022	115	6	101
pyruvic acid	2.5	2.5	0.2	6.2	98
	10	9.9	0.6	5.9	99
	25	25.3	0.9	3.7	101
	50	49	2	4	98
	100	104	7	6	104
fumaric acid	250	249	15	6	99
	0.25	0.23	0.03	12.05	92
	2.5	2.5	0.2	7.2	100
	10	10	1	10	101
	25	26	2	7	104
50	49	6	11	99	
100	97	11	12	97	
250	247	15	6	99	

## Results cont'd

Recovery, Matrix Effect, and Freeze/Thaw Stability of Select Organic Acids in Homogenates of Mouse Quadriceps

Organic Acid	Recovery (%)	Matrix Effect (%)
lactic acid	37	11
pyruvic acid	39	37
malic acid	28	30
fumaric acid	78	21

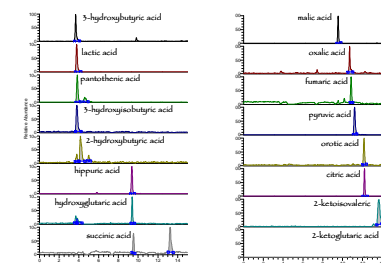
n = 3; recoveries could be improved with extractions in methanol instead of ethyl acetate

3-day Freeze/Thaw Stability of Quadriceps Homogenate

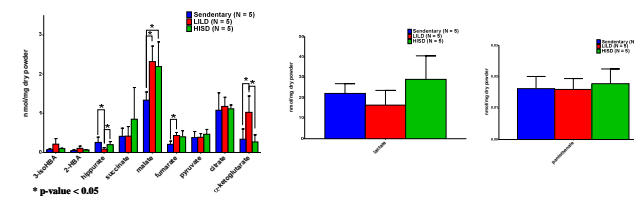
Organic Acid	Day 1 ( $\mu$ M)	S.D.	% CV	Day 3 ( $\mu$ M)	S.D.	% CV	% Stability
lactic acid	201.1	0.1	0.06	209	15	7	104
pyruvic acid	4.0	0.3	6	3.4	0.5	16	85
malic acid	15	1	9	14	2	14	93
fumaric acid	2.0	0.2	9	1.9	0.2	10	98

n = 3

## IC/MS of Organic Acids in Mouse Quadriceps



## Differential Profiles of Organic Acids in the Quadriceps of Exercised Mice by IC/MS



## Conclusions and Future Directions

- Ion chromatography is a complementary separation method to HPLC (*i.e.* ion-pairing and HILIC) to retain and quantitate polar, water soluble organic acids by mass spectrometry.
- Next Steps: Develop IC/MS methods for nucleotides, glycolysis, and pentose phosphate pathway intermediates.

## Acknowledgements

National Institutes of Health (U24 DK097209)

